

0040-4039(94)E0084-B

## Synthesis of 2(3H)-Imidazolethiones and 2(3H)-Imidazolones from β,γ-Alkynyl Carbanilides

Balachari Devan and Krishnamoorthy Rajagopalan

Department of Organic Chemistry, University of Madras, Guindy Campus, Madras - 600 025, India.

Abstract : B,  $\gamma$ -Acctylenic carbanilides cyclize and subsequently isomerize to 2(3H)-Imidazolethiones and 2(3H)-Imidazolones upon treatment with ammoniacal H<sub>2</sub>S and KOH in t-BuOH respectively.

Base catalysed intramolecular addition-cyclization strategy has been receiving a great deal of attention as a potential methodology for construction of five membered heterocycles.<sup>1,2</sup> A number of 2(3H)imidazolethiones have attracted considerable attention because of their pronounced antithyroid activity,<sup>3</sup> thus setting them apart as important synthetic targets. This prompted the communication of our findings viz., an expedient synthesis of 2(3H)-imidazolethiones and 2(3H)-imidazolones from compounds 1a-c.

N-propargyl carbanilide 1a was prepared in almost quantitative yield from carbanilide with propargyl bromide and  $K_2CO_3$  in acetone or acetonitrile.<sup>4</sup> Compound 1a was dissolved in methanol and the solution adjusted to pH=10 by adding NH<sub>4</sub>OH. H<sub>2</sub>S was passed at room temperature until a sample of the mixture no longer showed TLC spot corresponding to the starting material. The mixture was then poured onto ice and the solid which separated was filtered, dried and recrystallised from chloroform to afford 2(3H)-imidazolethione 4a (m.p: 228°C, CHCl<sub>3</sub>, 82% yield) (equation 1). The general applicability of this reaction was established with R=Cl and R=OCH<sub>3</sub>. Addition of few drops of 1N NaOH during the reaction enhances the rate and the reaction goes to completion in one hour.



Critical analysis of <sup>13</sup>C NMR data, appearance of an NH proton in <sup>1</sup>H NMR at  $\delta$  12.0 and failure to get the reduced product with NaBH<sub>4</sub> clearly showed the product to have structure 4. The reaction shows the initial attack of -SH on the nitrile carbon and cyclization occurs in a 5-exo-dig pathway leading to intermediate 3 which on subsequent isomerization affords compound 4. The presence of the NH proton was confirmed by D<sub>2</sub> O exchange in <sup>1</sup>H NMR.

Extension of this reaction to the B-allenyl system was investigated (eq2). The allenyl compound 5a was prepared following conventional methods.<sup>4,5</sup> This system failed to cyclize under conditions described as in equation 1, but resulted in the thio urea 6a in 70% yield.<sup>6</sup>



The drive for cyclization and the propensity to isomerization warrant further exploration of compound 1a-c. The cyclization of compounds 1a-c were also effected with powdered KOH in t-BuOH<sup>7</sup> at reflux temperature for 2 h. After work up, it afforded 2(3H)-imidazolones 7a-c as the only products in almost quantitative yield (Eq.3). All the compounds gave satisfactory spectral<sup>6</sup> and analytical data.

1 a, b B, c 
$$\frac{KOH, 1 - BuOH}{\Delta, 2h}$$
 o  $N$  (eq.3)

In conclusion, this study demonstrates that the 5-exo-dig cyclization operates both in ammoniacal H<sub>2</sub>S and KOH in t-BuOH thereby resulting in formation of 2(3H)-imidazolethiones and 2(3H)-imidazolone derivatives respectively under mild conditions and could therefore find useful applications in organic synthesis. Further work is in progress with compounds bearing good leaving group at the termini carbon of the alkyne in compound 1.

Acknowledgement: BD thanks UGC, New Delhi for the fellowship. We are thankful to Dr.B.Gopelan, Mr.S.Janardhanam and Dr.T.Rajamannar for useful discussions; Dr.S.P.Ganesh Raj (Zurich), Dr.R.Balasubramanian (University of Madras) and Head, RSIC, IIT, Madras for spectral data.

## References

3.

- Short, K.M.; Ziegler, C.B.Jr. Tetrahedron Lett., 1993, 34, 75-78. 1.
- 2. Marshall, J.A.; DuBay, W.J. J. Org. Chem., 1993, 58, 3435-3443.
  - Jackman, M.; Klenk, M.; Fishburn, B.; Tullar, B.F.; Archer, S. J. Am. Chem. Soc., 1948, 70, 2884. **a**.
    - Stanley, M.M.; Astwood, E.B. Endocrinology 1949, 44, 588-589. ь.
      - For a comprehensive review on 2(3H)-imidazolone and 2(3H)-imidazolethione, see Klaus Hofmann, "The c. Chemistry of Heterocyclic compounds" Imidazole and its derivative, part 1; Interscience Publishers, Inc., London. 1953, pp.63-91.
- Devan, B.; Rajagopalan, K. Synth. Commun., 1993, in Press. 4. 5.
  - Janardhanam, S.; Devan, B.; Rajagopalan, K. Tetrahedron. Lett., 1993, 34, 6761-6764. a.
  - Balakumar, A.; Janardhanam, S.; Rajagopalan, K. J. Org. Chem. 1993, 58, 5482-5487. b.
  - Dikshit, D.K. Ind. J. Chem., Sect. B., 1983. 22, 1144. c.

6. Spectral data of some selected compounds : 4h : Solid (m.p. 208°C, CHCl<sub>3</sub>) IR (KBr, v<sub>max</sub>, cm<sup>-1)</sup>; 1630, 1390, 1330, 1250, 1120, 1090, 1070. <sup>1</sup>H NMR (90 MHz, CDCl3), 6; 12.0 (br s, 1H,-NH) (disappears on exchange with D20), 7.42 (ABq, 4H, Ar-H), 6.55 (s, 1H, vinylic H), 2.19 (s,3H,-CH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz) δ : 160. 32 (s,C<sub>3</sub>), 136.09 (s), 133.95 (s), 129.17 (d), 127.07 (d), 125.95 (s), 115.32 (d), 9.81 (q). 6a : IR (CHCl<sub>3</sub>, υ<sub>max</sub>, cm<sup>-1</sup>); 3500, 3380, 1950, 1585, 1460, 1360. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ; 7.2 (ABq, 4H, Ar-H), 5.73-5.46 (br s, 2H, -NH<sub>2</sub>), 5.44-5.13 (m, 1H, "i=), 4.77-4.73 (m, 2H,  $\preccurlyeq_{H}^{H}$ ), 4.70-4.66 (m, 2H, -N-CH<sub>2</sub>), 2.39 (s-3H, Ar-CH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz) 209.44, 182.19, 139.02, 138.26, 130.78, 127.37, 85.76, 70.01, 54.34, 20.92. 7e : Solid (m.p.218°C, CHCl<sub>3</sub>-McOH), IR (CHCl<sub>3</sub>,  $\upsilon_{max}$ , cm<sup>-1</sup>); 3450, 1680, 1510, 1400 <sup>1</sup> H NMR (90 MHz, CDCl<sub>3</sub>)  $\delta$ ; 10.2 (br.s, 1H, NH) (exchange with D<sub>2</sub>O), 7.1 (ABq, 4H, Ar-H), 6.08 (s, 1H, vinylic H), 3.8 (s, 3H,-OCH<sub>3</sub>). 2.03 (s, 3H,-

CH<sub>3</sub>). <sup>13</sup>C NMR (DMSO-46, 22.5 MHz), 156.13, 152.07, 130.72, 121.89, 117.83, 113.88, 105.97, 55.04, 10.35.

7. Hall, J.H., Gisler, M., J. Org. Chem., 1976, 41, 3769-3770.

(Received in UK 11 November 1993; revised 21 December 1993; accepted 7 January 1994)